



MED iQ
presents



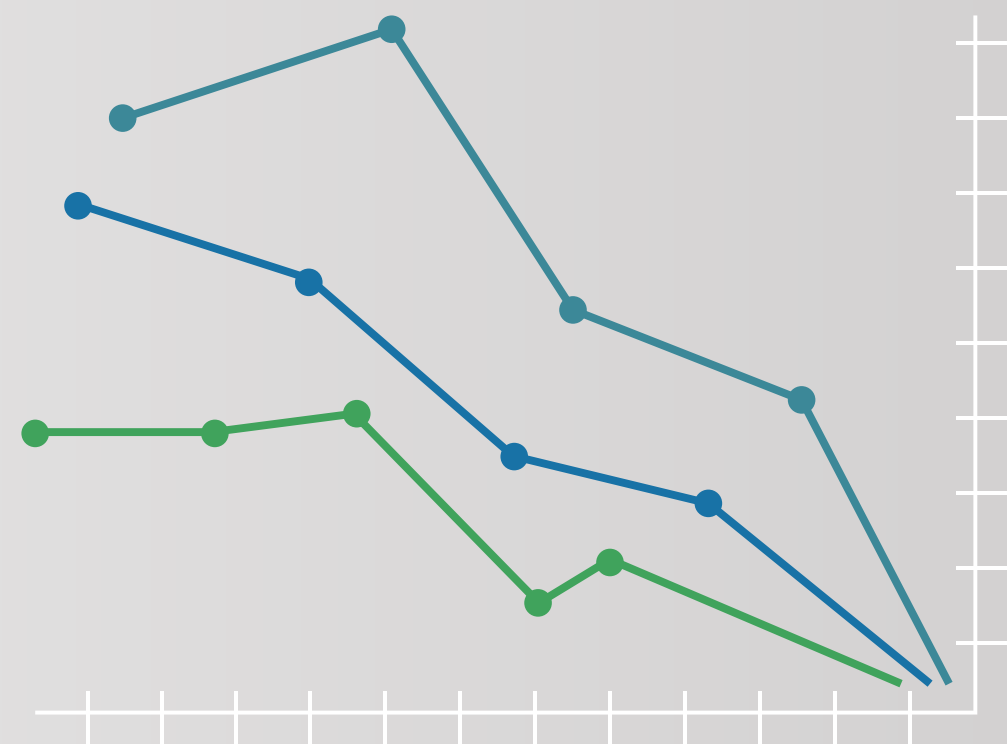
PHARMACOLOGY **DRUG PRINCIPLES**

– MOD 108 –

SCAN OR CLICK!



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Introduction

Henderson Hasselbalch

Weak acids: $pK_a = pH + \log \frac{\text{unionized}}{\text{ionized}}$

Weak bases: $pK_a = pH + \log \frac{\text{ionized}}{\text{unionized}}$

"Medical pharmacology" is often defined as the science of substances used to prevent, diagnose, and treat disease.

-Drugs: chemical substances stimulate or inhibit an existing cell function.

- Used for **treatment, prevention and diagnosis** of diseases or **modify a physiological process**.

1. Pharmacokinetics: It describes what the **body does to the drug** includes:

- Absorption, Distribution, Metabolism & Excretion [ADME]**.

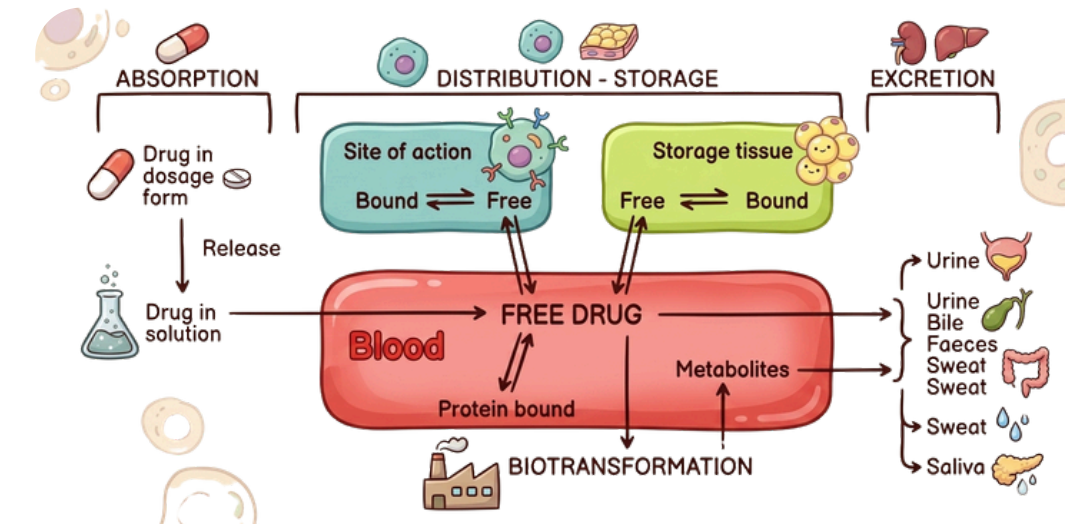
2. Pharmacodynamics: describes what drug does to the body, includes:

- Study of the pharmacological **actions of drugs**.
- Their possible **mechanisms of action**.

3. Pharmacotherapeutics: Study of **selection and use of drugs**

- For **diagnosis, prevention and treatment** of disease.

N.B The **main goal of a physician is to prescribe the most appropriate drug** for each individual patient **in adequate doses**, so that **therapeutic levels are obtained**, to **produce the desired response** with the **minimal adverse effects**.



N.B: Lipid/water partition coefficient: ratio of drug conc in lipid medium to its conc in water medium when it is put in lipid water system.

Passage of drugs across cell membranes

- The cell membrane is lipid sheet interrupted by protein in form of:
 - Ion channels, carriers and receptors, and contains water filled channels.**
- Passage** of drugs across cell membranes occurs mainly by:

1. Simple diffusion

2. Carrier mediated transport

2) Carrier mediated transport:

chief process involved in absorption & distribution

1) Simple diffusion

Depends on:

1) Concentration gradient:

- From one side of cell membrane **to the other along conc gradient**

2) Molecular size: The **smaller** the molecular size **the better** absorption.

3) ↑ Lipid solubility & lipid/water partition coefficient, the **better** absorption.

4) Ionization degree:

- ↑ **Ionization**, the ↓ **lipid solubility**.
- Degree of ionization **depends on:** **pKa of the drug & pH of the medium**.

5) No energy & no carriers are needed.

- pKa (ionization constant of a drug).

- pH at which **½ of the drug is ionized and ½ non-ionized**
e.g. pKa of aspirin 3.5, So in pH 3.5 aspirin will be 50% ionized and 50% non-ionized.

- pH of the medium.

1. Most drugs are either weak acids or weak bases.

2. Weak acids: absorbed in acidity as it's ↓ ionized so ↑ lipid soluble.

3. Weak bases: absorbed in alkalinity as it's ↓ ionized so ↑ lipid soluble.



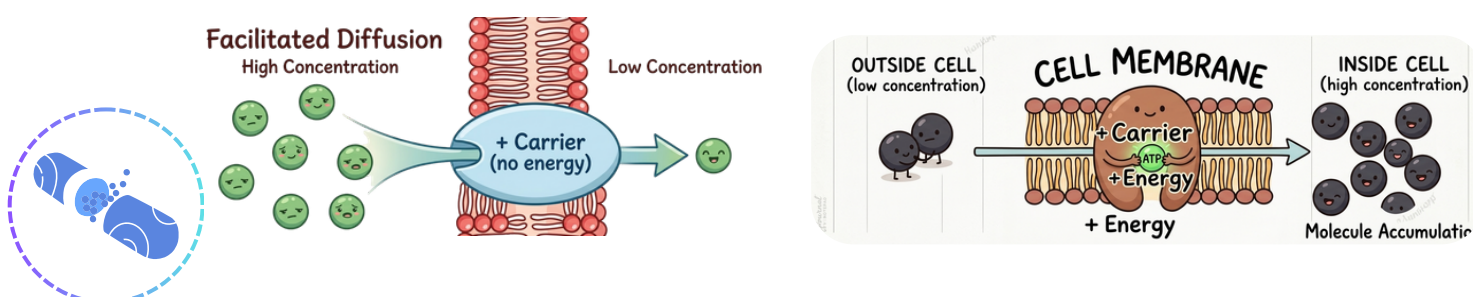
- Acidic drugs as aspirin ↓ ionized in stomach acidity → ↑ absorption in stomach.

- When filtered by the glomeruli, **big aspirin %** can be **reabsorbed** from the renal tubules **since urine is usually acidic**.

- So **making urine less acidic** (more alkaline) would **↑ excretion of aspirin** because **less aspirin will be reabsorbed**.

- Weak base drugs as amphetamine will be **better absorbed from the intestine & acidic urine will ↑ excretion**.

a) Facilitated diffusion	b) Active transport:
Drugs are transferred across cell membranes: Along their concentration gradient. Carrier for substances which are too large or lipid insoluble to diffuse passively No energy is required. e.g. glucose uptake by cells.	Drugs are transferred across cell membranes: Against their concentration gradient - Transport requires a carrier - Transport requires energy e.g. secretion of penicillin by renal tubules.



Pharmacokinetics

ADME

Absorption

Transfer from site of administration → systemic blood

Drug related



Nature

• Inorganic > Organic



Solubility



- Drug MUST be **Water & lipid soluble**.
- Must be completely dissolved in H₂O to be absorbed.
- ↑ **lipid solubility**
- ↑ **Lipid/Water partition coefficient**
- ↑ **absorption**.

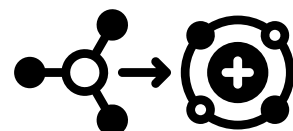
Valency

• Ferrous > Ferric



Ionization

- **Non-ionized**
- **More lipid soluble**
- **Better absorption**

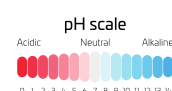


Pharmaceutical Preparation

• **Solution > Suspension > Tablet.**



Gut pH



- Weak **acids** better absorbed in **acidic** media.
- **(aspirin) (stomach)**
- Weak **bases** better absorbed in **basic** media.
- **(amphetamine, ephedrine) (intestine)**

Specific factors

- **Intrinsic factor** for Vit B12.



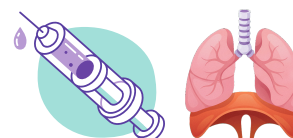
Distribution

Patient related



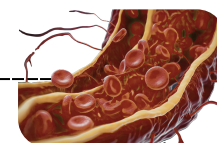
Route of Administration

• I.V. & inhalation > I.M > S.C > Oral > Skin



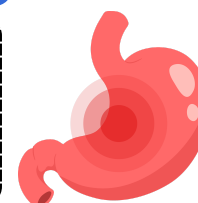
Absorbing Surface

- Vascularity:
 - **Alveoli > Skeletal muscle > SC tissues.**
- Surface area:
 - **Alveoli > Intestine > Stomach** (Intestine has 10³ times the area of stomach) ; due to microvilli & rich blood.



Absorbing surface State of health

- Oral **absorption is DECREASED**:
 - Gastritis
 - malabsorption
 - Diarrhea



Gut contents

Other drugs & food presence

- **Adrenaline S.C → V.C**:
 - ↓ **Absorption of Local anesthetics**
 - ↑ **duration** of action of local anesthetics.
- **Milk (Calcium)**: ↓ **Oral abs Tetracyclines** (Abs)
- **Grape fruit juice**: ↑ **absorption of drugs**
 - by **inhibiting P. glycoprotein** (cause **reversed transport** from gut wall to lumen).
- **Tea**: ↓ **iron absorption** ; due to **tannic acid**.
- **Food** presence:
 - **Bad** → **dilutes** Drugs & **compete** for absorption.
 - **Good** → with **IRRITANT** drugs e.g. aspirin & iron.

Gut motility & rate of dissolution

- **Prokinetic drugs increase motility.**
 - e.g. **Metoclopramide**
- **Atropine** **inhibits** motility.

Systemic circulation

- **Absorption is DECREASED**:
 - Shock
 - Heart failure



First pass effect

drug metabolism before reaching systemic circulation.

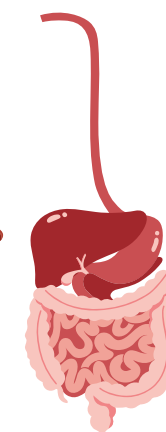
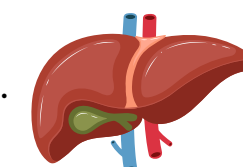
- Gut first pass effect:

- **Gastric acidity**: destroys **benzyl penicillin**.
- **Digestive enzymes**: destroy **insulin**
- **Mucosal enzymes**: destroy **chlorpromazine**.



- Hepatic first pass effect:

- Drugs **extensively metabolized**: **Nitroglycerine**
 - **To avoid**: change the route of administration.
 - Nitroglycerine is **administered sublingually**.
- Drugs **metabolized to large extent**: **propranolol & nitroglycerine**.
 - **To compensate**: **INCREASE** the oral dose.



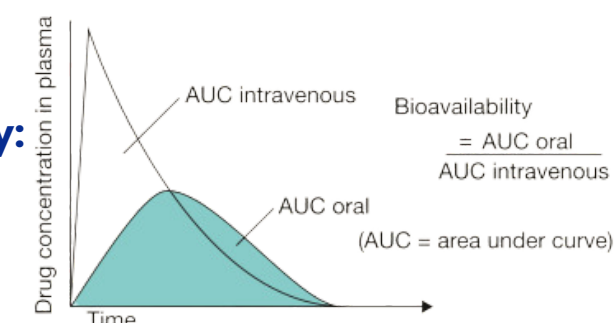
Bioavailability

% unchanged drug reaches systemic circulation after administration.

- **100 % after IV administration.**
- Variable after oral administration.

- Factors affecting oral bioavailability:

1. Amount of drug absorbed = Factors affecting GIT absorption.
2. First pass metabolism.



Pharmacokinetics

Physicochemical properties

Molecular weight & degree of ionization & lipid solubility.

Plasma proteins binding

drugs may **bound to plasma proteins (chiefly albumin)**.
Salicylates are **strongly bound** to plasma proteins.
So, drugs are carried in blood in 2 forms:

Free form	Bound form
active diffusible	Inactive non-diffusible
metabolized	not metabolized
excreted	not excreted (act as reservoir)

Amount of drug bound change according to:

- Affinity for binding sites:

Competition between drugs, may displace other.

- e.g. **Salicylates** can **displace warfarin** → **hemorrhage**.

- Hypoalbuminemia → ↑ free drug:

- e.g. in **starvation**, **malnutrition**.
- Therapeutic dose changes to toxic dose** e.g. **Phenytoin**.
- The ↑ binding, the ↑ duration** e.g. **sulphonamides**.

Passage across barriers

Through breast milk:

- Most drugs for lactating women detectable in milk.
- milk pH is more acidic (7.0)** than that of plasma (7.4), so **basic drugs ionize & accumulate in milk (ion trapping)**
- Milk contains more fat than plasma which favors retention of lipid soluble drugs.

Passage to fetus across placental barrier:

placental barrier acts like a cell membrane.

So **non-ionized, lipid soluble** drugs **pass** from the mother **to fetus more easily**
Drugs passing may cause: **During pregnancy** → **Teratogenicity (Tetracyclines)**.
During Labor → **Neonatal asphyxia (Morphine)**.

Distribution

drug distributed among body compartments after absorption



One compartment model (IV)

Drugs having **↑ MWT are trapped in IV compartment** distributed in 4 L which's = 6% of a 70kg BW.

- e.g. **heparin** or drugs highly bound to proteins (**warfarin**).

Two compartments model (extracellular)

The **IV space** and **ISF** space.

↓ **MWT, not lipid soluble** are distributed to 14 L = 20% BW

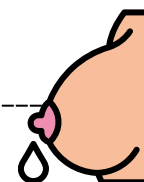
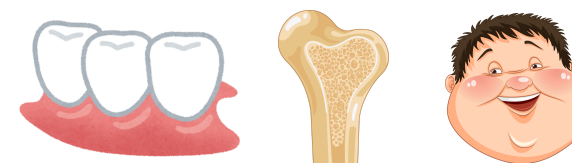
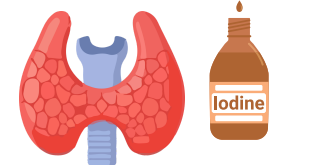
- e.g. **quaternary ammonium** & **aminoglycoside** antibiotics.

Multicompartment model (Extra & intracellular)

Drugs are distributed **to total body fluids (42L / 70 Kg)**. They are of ↓ **MWT & hydrophobic (Lipid soluble)**.

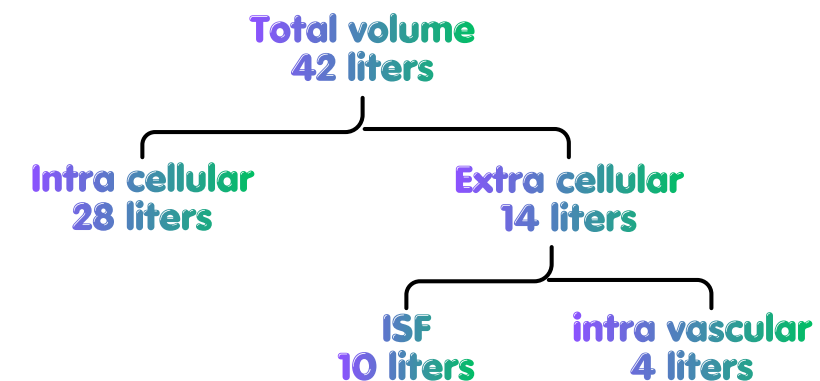
Special affinity for certain tissue

- Iodine in thyroid.**
- Thiopentone in fat.**
- Tetracycline and Ca in bone & teeth.**



LOADING ...

first year



- Apparent volume of distribution (Vd):

Assuming that body consists of 1 big fluid compartment and that total amount of drug (A) would be uniformly distributed in that compartment such that it would have the same concentration as in plasma (C).

$$Vd = \frac{\text{dose (mg)}}{\text{plasma conc (mg/L)}}$$

Significance:

1. Rough estimation of drug distribution:

- ↓ **Vd** drugs retained in vascular compartment
- ↑ **molecular weight & ↑ plasma proteins binding**
- ↑ **Vd** drugs occupy multicompartment or concentrated in tissues
- e.g. **digoxin** in heart.
- It is **called apparent Vd** because Vd of some drugs as **digoxin (500L/70kg)** can **extremely exceed any physical volume in the body**.

2. Determine loading dose & total drug amount in body:

3. In cases of drug toxicity

- **Dialysis** can be done **only** with ↓ **Vd** which means that most of **drug present in circulation**.
- e.g. **Aspirin**.



CNS across Blood Brain Barrier (BBB)

- BBB composed of basement membrane & 3 cellular elements: endothelium, pericytes, astrocyte end feet
- Aim: shield brain from harmful substances.
- **Non-ionized, lipid soluble** drugs **can pass the BBB**.
- e.g. **Physostigmine** stimulate the CNS, **neostigmine** can **not**.
- **Inflammation (Meningitis)** ↑ **permeability of B.B.B.**
- e.g. **Penicillin** can **pass inflamed meninges** but **NOT normal one**.

Pharmacokinetics

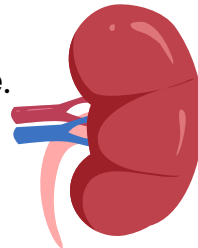
ADME

Excretion

- Removal of a drug from the body
- **Most important route via kidney** into urine.
- Other routes: GIT, skin glands, and lungs.

Renal

1. **Nonvolatile** drugs and **metabolites**.
2. Renal excretion is the **result of glomerular filtration and active tubular secretion & reabsorption**.
3. **Passive Glomerular filtration** for:
 - **Water soluble**
 - **Nonbounded** drugs with M.W. < 500 daltons.
4. **Active Tubular Excretion**:
 - **Saturable & Site for competition & Drug Interaction**.
5. **Changes in urinary pH**:
 - **Alkalinization**:
 - By giving **Na or K Acetate, Bicarbonate or Citrate**.
 - **Acidification**:
 - By giving **NH₄Cl or ascorbic acid (Vit C)**.
6. The **clearance of some drugs depends mainly on renal excretion (Little or no metabolism)**
 - **BE CAUTIOUS in Renal patients.**



Site of Metabolism (Organs):

- **Liver is the main site** for biotransformation
- The **Lungs** have enzymes that can metabolize:
 - **angiotensin I, prostaglandins & nicotine.**
- In the **presence of UV light**:
 - **Skin** converts **provitamin D to pre-vitamin D**.
- The **kidneys** convert **vitamin D to a more active form**.
- **Plasma choline esterases inactivate acetylcholine & other similar compounds.**
- **G.I.T. & Gut flora** metabolize **Tyramine & Histamine**

Phase-I (Non-Synthetic) Oxidation, Reduction & Hydrolysis

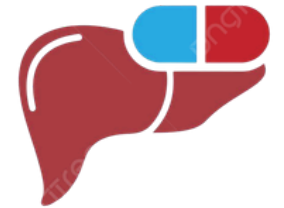
1. **Oxidation**:
 - **Alcohol** → Acetaldehyde → Acetic acid → CO₂+H₂O+E
2. **Reduction**:
 - **Chloral hydrate (Active)** → Trichloroethanol (More **active**)
3. **Hydrolysis**:
 - **Acetylcholine (Active)** → Acetic acid + Choline (Inactive)

Phase-II (Synthetic) Conjugation

- Usually leads to inactivation
- **Glucuronidation of paracetamol** → inactivation
- **Acetylation of isoniazid** → inactivation
- **Glycine conjugation of aspirin** → inactivation
- May lead to activation
 - **Methylation of adrenaline** → **active noradrenaline**
 - **Morphine** → **Morphine-6-glucuronide** (More **active**)

active lipid soluble → **inactive water soluble**
 to be easily excreted.

Metabolism



- Results of Phase-I Metabolism:

1. Inactivation (commonest fate):

Adrenaline & Noradrenaline → **Vanil Mandilic Acid (VMA)**

2. Activation:

Inactive (Prodrug) → Active metabolite

CortisONEE → **Hydrocortisone (CortisOOLL)**

3. Maintain Activation:

Active Drug → Active Metabolite

Diazepam → **Nor-diazepam**

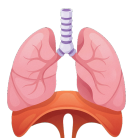
4. Toxication:

- **Methyl alc** → **Formaldehyde** → **Blindness**



N.B Most drugs undergo phase I then Phase II.
 • Some have reverse order e.g. **Isoniazid** Acetylation (Phase II) then → Hydrolysis (Phase I).

The Lungs



- **Gases (CO₂).**
- **Volatile Liquids.**
- **Halothane** that is used in **anesthesia**.

Skin Glands

1. **Sweat**:
Vit B-1, Hg, As, rifampicin
2. **Milk**: affect suckling baby e.g. **purgatives**



N.B Rifampicin which causes **sweat red discoloration**.

The Alimentary Tract

1. **Saliva (pH=8): Iodide.**
2. **Stomach: Morphine.**
3. **Bile:** some may be secreted in bile by mechanism similar to renal tubular secretion. Such drugs reach the intestines in the bile & have one or more of the following fates:
 - Be excreted in large intestine.
 - Be absorbed again from the small intestines **and undergo enterohepatic circulation.**
 - e.g. **Rifampicin.**
4. **Large Intestine:**
via the bile or unabsorbed oral drugs.



	HME	Non-Microsomal enzymes
Site	Smooth endoplasmic reticulum	Cytoplasm, Mitochondria, etc.
Organs	Mainly Hepatic	All organs
Phase-I	Oxidation / Reduction (CYT P-450)	Oxidation/Reduction & Hydrolysis
Phase-II	Glucuronidation ONLY	All Except Glucuronic acid
Induction	Inducible	not Inducible
Substrate	Usually lipophilic	Lipophilic & hydrophilic



Pharmacodynamics



Possible Mechanism of Drug Action:

Drugs may act through one or more of the following

A | Physical

- **Absorption:** eg;
 - Kaolin in Diarrhea
- **Osmotic:** eg;
 - MgSO₄ as saline purgative
 - Mannitol as osmotic Diuretic

C | Interference with Cell Division

- eg; Anti Cancer Drugs

E | Inhibition of important Enzymes

- Physostigmine & Cholinesterase
- Aminophylline inhibits Phosphodiesterase (PDE)
- Aspirin & Cyclooxygenase (COX)

G | Inhibiting Membrane Carriers

- Thiazide Diuretics block NaCl Symport in distal tubules of kidney → ↑ Na & water excretion

B | Chemical

- **Neutralization:** eg;
 - NaHCO₃ (Antacid) + HCl in treatment of Hyperacidity
- **Chelation:** organic compound + Heavy metal → produce nontoxic easily excreted complex. eg;
 - Desferrioxamine for Fe³⁺

D | Interference with a Metabolic Pathway

- eg; Sulfonamides compete with PABA in bacteria → inhibition of formation of Folic Acid

F | Action on Membrane Ionic Channels

- **Local Anesthetics** as procaine block Na⁺ Channels → Stabilize nerve cell membrane

H | Action on a Specific Receptor

- Most common mechanism

Definition of a Receptor:

A receptor is a **chemo-sensitive & selective** protein that interacts with a **ligand** (drug, transmitter or hormone) to produce a **response**. The receptor may be located in the membrane, cytoplasm or nucleus.



K_a = Association constant with the receptor

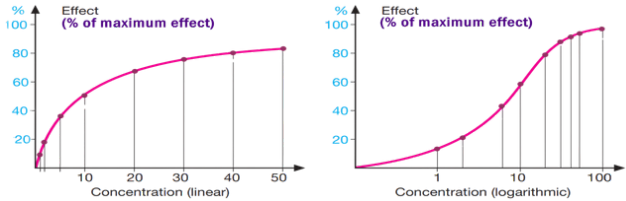
K_d = Dissociation constant for the receptor

Affinity: ability of the drug to fit onto receptor to form Drug-Receptor (D/R) Receptor

Efficacy / Intrinsic Activity (IA) = Ability of D/R Complex to evoke a response

Relationship between drug concentration and response:

Represented by **Concentration-Response curves**, usually **hyperbolic** but becomes **sigmoid** when **log** concentration is plotted against response.



Concentration-Response Curve of Drugs:

Relation between Log Conc. and Response (effect) of drug.

Useful to know **Effects:**

- Minimal Effect (E_{min})
- Maximal Effect (E_{max})
- Submaximal Effect (E₅₀ = 50% effect)

Useful to know **Doses:**

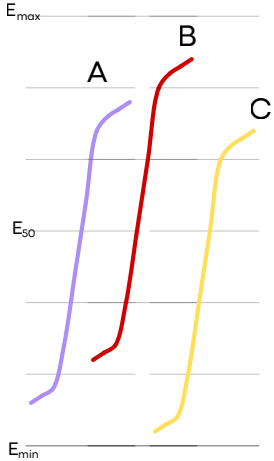
- Produce Minimal Effect (ED_{min})
- Produce Maximal Effect (ED_{max})
- Submaximal Effect (ED₅₀ → E₅₀)

Useful to **Compare** drugs:

- **Efficacy** → Compare E_{max} (B>A>C)
- **Potency** → Compare doses that produce same submaximal effect (A>B>C)

Useful to determine **type of blocker** if:

- **Competitive** → Parallel shift to right (↓ Potency) with same E_{max} (Efficacy)
- **Non-competitive** → Non-parallel shift to right (↓ Potency) with ↓ E_{max} (Efficacy)



Types of Inhibitors:

A | Competitive Antagonists

- Antagonists bind **Reversibly** with the receptors
- Antagonists can be **Displaced** by excess agonist → **(Surmountable Block)**
- **Parallel** shift of the Conc-Reponse curve of the agonist to **RIGHT** → ↓ **Potency**
- **No effect** on maximum response (E_{max}) = **Same Efficacy**

Examples:

→ Propranolol → Atropine → Naloxone

B | Noncompetitive Antagonists

- Antagonists **NOT Displaced** by agonist → **(Nonsurmountable Block)**
- **Non-Parallel** shift of the Conc-Reponse curve of agonist to **RIGHT** → ↓ **Potency**
- **Decrease** maximum response (E_{max}) to agonist → **Decreased Efficacy**

Types of Ligands:

Drugs are classified, according to the nature of their interaction with receptors in **Agonists**, **Partial Agonists**, and **Antagonists**.

Drug	Definition	Affinity	Efficacy	K _a & K _d	Effect	Example
Agonist	Drugs that have affinity , efficacy , and rapid disassociation	High	High	Rapid	Full response	Acetylcholine Adrenaline
Partial Agonist	Produce lower response at full receptor capacity than agonists Act like weak agonist in absence of agonist OR like antagonist in presence of agonist		Low	Slow / Moderate	Full response OR Block agonist	Pindolol Sumatriptan
Antagonist	Drugs that block receptors. They have affinity, no efficacy & slow disassociation		NONE	Slow	Block Agonist	Atropine

Types of Noncompetitive Block

Reversible	Irreversible
Antagonists bind REVERSIBLY to receptor	Antagonists bind IRREVERSIBLY
Block ends by: Metabolism of Antagonist	Block ends by: Resynthesis of Receptors
Usually short duration of action	Usually long duration of action
eg; Nicotine LD & Succinylcholine	eg; Phenoxybenzamine & Organophosphorus

N.B: Chronic use of drugs affects the number & Sensitivity of Receptors

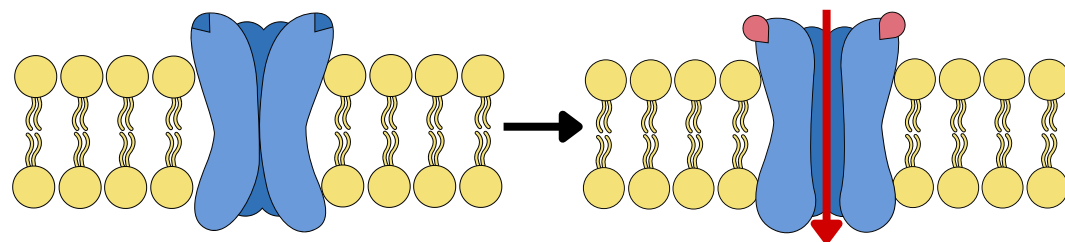
Long use of **Agonists** → Decrease in number & Sensitivity of Receptors → **Down Regulation**

Long use of **Antagonists** or **Transmission inhibiting Drugs** → Increase number & Sensitivity of Receptors → **Up Regulations**

Ligand-Gated Ion Channels:

Ligand binds to the extracellular domain of a ligand-gated ion channels

- **Membrane-bound** receptors located on the **gate** of an ion channel
- Binding of **Ligand** to **Receptor** results in conformational change to receptor, resulting in **change** of membrane **permeability** to ions.
- **Milliseconds** Between **binding** and **response**
- **Examples:**
 - **A.Ch** + Nicotinic Receptors → ↑ Na⁺ Influx → Depolarization
 - **GABA** + GABAA Receptors → ↑ Cl⁻ Influx → Hyperpolarization



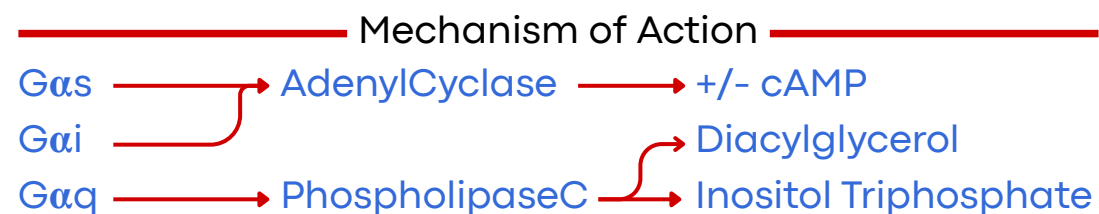
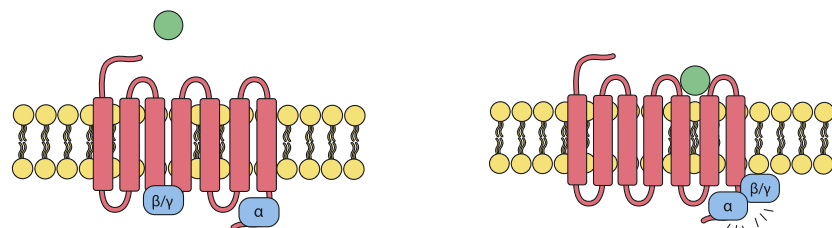
Types of Receptors

- Receptors have 2 main functions:
 - Ligand Binding
 - Generating Effector Response by interacting with closely associated cellular proteins called "Signaling Systems"

G. Protein Coupled Receptors & 2nd Messengers:

Ligand binds to domain of transmembrane receptor coupled to a G Protein

- **Cell surface receptors** which facilitate binding of **GTP** to specific proteins on cytoplasmic surface (**G Protein**)
- **G. Protein** regulates activity of membrane **enzymes** & ion **channels**.
- **G. Protein** is a membrane protein made up of 3 Subunits (**α, β, γ**)
- **G. Proteins** **EITHER** regulate intracellular 2nd messages (**α**-subunits with GTPase activity)
- **OR** control opening of ion channel (**β**-subunits & **γ**-subunits)
- There are different **Gα** subunits (**Gα_s** - **Gα_i** - **Gα_q**)
- **Few Seconds/Minutes** between ligand bind & effect.

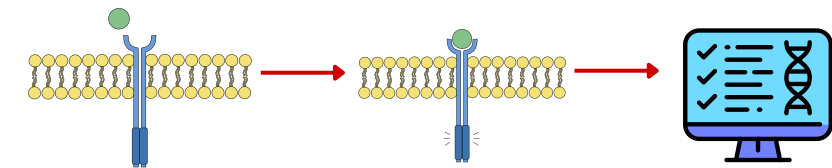


- Examples**
- **Adrenaline** + **β** Receptors → **Gα_s** Protein → stimulate **A.C** → ↑ cAMP
 - **Adrenaline** + **α₂** Receptors → **Gα_i** Protein → inhibit **A.C** → ↓ cAMP
 - **A.Ch** + **M1&3** Receptors → **Gα_q** Protein → stimulate **PhospholipaseC** → ↑ IP₃ & DAG → Ca²⁺ + Calmodulin → Response
 - **A.Ch** + **M2** Receptors → **G_{βγ}** → K⁺ Efflux → Hyperpolarization

Enzyme (Tyrosine Kinase) Linked Receptors

Ligand Binds to ECM Domain of a receptor → **activates kinase**

- **Polypeptide Receptors** consisting of **Extracellular** domain (Where Insulin / Growth Hormone attach)
- Connected to **Cytoplasmic** Enzymatic domain containing **Tyrosine Kinase** Enzyme, which when activated phosphorylates:
 - 1 Signal Transducer → Non-genomic Actions
 - i Takes **seconds/minutes**; eg: Hypoglycemia of Insulin
 - 2 Activate transcript → Separate from receptor → Cross nuclear membrane → Gene transcription → Genomic Action
 - i Takes **Hours**; eg: anabolic effect of insulin & GH



Gene-Active Intracellular Receptors (DNA Linked Receptors)

Lipid soluble ligand diffuse across membrane to modulate gene expression

- **Cytosolic or Nuclear** Receptors that modulate transcription of **genes** leading to changes in **protein synthesis**
- ★ **eg**; Lipid soluble ligand (Steroid hormones, Thyroxine, Vit. D) + Intracellular Receptor → **DNA Transcription** → **mRNA** → change in **Protein Synthesis**
- **Delayed Lasting** Effect (**Hours** between ligand binding and cellular response)
- **Drugs act on more than one receptor**
eg; Acetylcholine



Type A: Augmented or Predictable Undesirable Adverse Effects

Predictable undesirable effects related to normal pharmacological actions of the drug.

A | Side Effect

- An **unintended & inevitable** expected effect of drug related to its **pharmacological** properties & is produced by the **Therapeutic Dose** of drug.
- Usually **Unpleasant** or **Hurtful**
- **Examples:**
 - Dry mouth from **atropine** when used as antispasmodic
 - Gastritis induced by **NSAIDs** when used as **antipyretic**

B | Secondary Effect

• **Indirect consequence** of a drug related to its pharmacologic properties (**Pharmacodynamics**) and produced by **Therapeutic dose**. **Examples:**

Drug	Primary Effect	Secondary Effect
Oral Broad Spectrum antibiotics. Used to treat Resp. Tract infections	Killing pathogenic bacteria in the Respiratory tract	• $\downarrow$ Synthesis Vit B & K • Overgrowth of resistant bacteria $\rightarrow$ Super infection
Inhaled corticosteroids used to treat bronchial asthma	Anti-inflammatory & Anti-allergic effect on bronchi	Suppress cell-mediated immunity $\rightarrow$ overgrowth of oral bacteria $\rightarrow$ Oral Candidiasis

C | Overdose

- **Exaggerated** pharmacological action of the drug occurs when drug attains high **toxic** plasma levels.
- **Causes:**
 - **Very large** dose of drug / **larger** than therapeutic dose in drug with **narrow margins**
 - **Drug accumulation:** on repeated administration with impaired elimination (Liver/Kidney Failure)
- **Example:**
 - Insulin Large Dose $\rightarrow$ Hypoglycemia

D | Supersensitivity

- **Exaggerated** pharmacological action of the drug in response to **small therapeutic dose** in **SOME** individuals due to tissue **hyper-responsiveness**.
- **Management** $\rightarrow$ Decrease dose
- **Example:**
 - **Hyperthyroidism** $\rightarrow$ Sensitive to **sympathomimetics**

E | intolerance

- **A Severe toxic effect in response to therapeutic doses.**
- **Example:**
 - **Statin Intolerance** $\rightarrow$ Severe liver & muscle injury

F | Interactions

- See **Drug Interactions Mindmap**

G | Direct Effects

- **Drug** or its **metabolites** react with tissues $\rightarrow$ **damage cells**.
- **Cytotoxicity** $\neq$ immune mediated
- **Examples**
 - **Cardiotoxicity:**
 - Halothane & Doxorubicin
 - **Hepatotoxicity:**
 - Halothane & Paracetamol
 - **Nephrotoxicity:**
 - NSAIDs & Aminoglycosides
 - **Neurotoxicity:**
 - Aminoglycosides $\rightarrow$ 8th cranial N
 - **Bone Marrow Inhibition:**
 - Chloramphenicol

Type B: Bizzare / Unpredictable Adverse Effects

Unpredictable undesirable effects related to pharmacological actions of the drug.

A | Idiosyncrasy

- **Unpredictable abnormal** response due to **genetic abnormality**.
- Occurs at **first exposure**
- **Examples:**
 - **Hemolytic anemia** in patients with **favism** (G6PD Deficiency)
 - Induced by **Primaquine**, **Aspirin**, and **Sulfonamides**.

B | Allergy (Hypersensitivity)

- **Unpredictable Abnormal** response to drugs due to immune reactions.
- **Drug** or its **Metabolites** may act as **antigen** or a **hapten**.
- **Hapten:** molecule that cant cause an allergic reaction by itself but causes an allergic reaction when it binds to **another carrier molecule**.
- **Cross Allergy** can happen between **chemically related** drugs (eg; Penicillins & Cephalosporins)
- Sometimes **Excipient** is the cause of allergy.
- Doesn't happen on **First Exposure** to drug (**IgE Mediated Only**)
- **IgE Mediated Reactions** may lead to drug resistance.
- **Management:** Don't reuse same or related drug again.

Type D: Delayed Effects

Delayed effects appearing after long use, even after stopping drug.

A | Teratogenicity (Dysmorphogenesis)

- **Teratogen** is any agent that results in abnormalities in a fetus, or a child after birth, as result of maternal exposure.
- Highest risk during **First Trimester**.
- **Drug use during pregnancy also cause:**
 - Abortion, growth retardation, and pharmacological effects in neonates that are reversible.
- **Examples:**
 - **Thalidomide** $\rightarrow$ **phocomelia** (Malformed short limbs)

B | Mutagenicity

- Induction of **gene mutations** in Somatic cells or germ cells.
 - **In Somatic Cells** $\rightarrow$ **Carcinogenicity**; eg: Sirolimus & Risk of skin cancer
 - **In Germ Cells** $\rightarrow$ **Infertility** or Teratogenicity

Type E: End of Use Effects

Adverse Effects after sudden stopping of drug after long use.

- **Abstinence Syndromes** of addicting drugs as **morphine**, **cocaine**, **barbiturate**.
- **Acute Addisonian Crisis** of chronic **Steroid** Therapy.
- **Worsening of Existing Disease:** • Myocardial infarction after beta blockers • Rebound Hypertension & clonidine • Withdrawal seizures & anti-epileptics • Rebound Insomnia & hypnotics • Rebound nasal congestion & Decongestants

Type C: Chronic Effects

Effects of prolonged / continuous use of drug.

A | Tolerance

- **Lack** of drug sensitivity that responds to dose **>** therapeutic dose
- **2 Types:** **Congenital** (Inborn / innate) & **Acquired**

————— **Congenital** —————

- **Racial:** Black races tolerate mydriatic effect of ephedrine
- **Species:** Some herbivores tolerate ingestion of large amounts of **Atropine** without toxicity (unlike humans). This **MAY** be due to strong **atropinase**.
- **Inter-individual Variability:** Pharmacogenetics
 - Some humans need higher doses to achieve effect.
 - **Possible Explanations:**
 - **Kinetics:** $\downarrow$ Drug absorption or $\uparrow$ Elimination
 - **Dynamics:** $\downarrow$ Number of receptors / $\downarrow$ post-receptor signaling.
 - **Examples:**
 - Congenital **CYP2C19** Deficiency $\rightarrow$ No activation of **clopidogrel** $\rightarrow$ **Triple** maintenance dose to **225mg** Daily.
 - Genetic mutation of **vitamin K epoxide reductase** $\rightarrow$ **Reduced** affinity to **warfarin** $\rightarrow$ requires doses **5x-20x** higher than average.

————— **Acquired** —————

- **Progressive** decrease of drug sensitivity due to **continued administration**.
- **Types:**
 - **Cross-Tolerance:** between similar drugs; eg: different types of nitrates.
 - **Tachyphylaxis (Acute Acquired Tolerance):** Rapid tolerance following a few doses over a short period.
- **Mechanisms of Acquired Tolerances:** Either due to **Kinetics** or **Dynamics**.
 - **Kinetic Changes:** $\downarrow$ Absorbance or **HME Inducers**
 - **Dynamic Changes:**
 - **Down-regulation** of receptors; eg: β_2 or Opioid receptors
 - **Antibody formation;** eg: Insulin

————— **Notes** —————

- **CYP2C19** is a liver enzyme that is responsible for metabolism of 10% of all known drugs. it is an example of an **HME Inducer**.

B | Drug Dependence

- **3 Types:** **Habitual/Psychic Dependence**, **Physical Dependence**, and **Addictions**.

————— **Habitual / Psychic Dependence** —————

- The **Drive** for continuous administration of the drug to produce **pleasure**, sense of **well-being**, and **enhance cognitive functions** (attention & working memory)
- **Examples:** **Coffee / Tea, Smoking** (Now considered Nicotine Addiction)
- **Symptoms:** Lasting for **Short** period of time.
 - Anxiety & Depression • Dysphoria • Deteriorating Cognitive Function • Craving

————— **Physical Dependence** —————

- A state of body **Adaption** to drug presence $\rightarrow$ Abrupt stop of administration causes drug-dependent **withdrawal syndrome**.
- **Examples:** **Opioids** like morphine & **Amphetamines**.
- **Symptoms:** Depends on the drug.
 - **Opioids:** Muscle pain, anxiety, insomnia, lacrimation (increased tears), rhinorea, sweating, diarrhea, dilated pupil, and high blood pressure.
 - **Amphetamines:** Sleepiness, fatigue, increased appetite, depression, cognitive decline, aches and pains.

————— **Addiction / Substance Dependence** —————

- **Neuropsychological** disorder caused by repeated use of a substance and consisting of a strong internal drive to use that substance.
- **Essential Features:**
 - **Impaired control** over substance use
 - **Craving** (Strong desire to use substance) often accompanies impaired control.
 - **Increased Priority** given to substance use over other aspects of life.
 - **Tolerance, Physical dependence, and Withdrawal** may be expected, however **impaired control** or **increased priority** should be present.

-- C -- Latrogenic Diseases

- **Any** drug induced disorder that mimics a natural disease.
- **Example:** **Drug Induced Cushing Syndrome**
 - Long-term use of **cortisol** $\rightarrow$ **Latrogenic Cushing Syndrome**
 - **Reversible** after drug discontinuation



Drug Interactions



Overview

Drug Interactions are altered pharmacological responses due to multiple drugs acting concurrently

BENEFICIAL (Desired)
Obtained by combining drugs with different mechanisms or drugs that correct adverse effects of each other

Harmful (Undesired)
Can be predictable effect of one or both drugs (Type A reaction) or an unpredictable toxicity (Type B).

Types of Drug Interactions

1 Pharmaceutical (Outside Body)

Physical or chemical interactions between drugs before administration (in Vitro)
Example: mixing ceftriaxone (an antibiotic) to IV containing calcium (eg: Ringer's lactate) forms a precipitate

2 Pharmacokinetic (Affects ADME)

Affects the Absorption, Distribution, Metabolism (Biotransformation), and Excretion of drugs

3 Pharmacodynamic (Affects Drug Response)

Drug interactions occur at sites of action, receptor sites or secondary physiological mechanisms; leading to changes in drug responses.

Results of Drug Interactions	
Addition / Summation (1 + 1 = 2)	Resultant action equals sum of individual drug actions; eg: Aspirin + Paracetamol
Synergism (1 + 1 > 2)	Resultant action exceeds sum of individual drug actions; eg: Cotrimoxazole (trimethoprim + sulfamethoxazole)
Potentialiation (1 + 0 > 1)	One drug has no action but increases effect of another drug (>1) eg: Aspirin + Barbiturates
Reversal of Action	Adrenaline after Phenatolamine (α Blocker) → Hypotension
Antagonism	Occurs when using drugs of opposing actions

Pharmacokinetic Drug Interactions

- Drugs affecting gut motility:** Motility (↑ by metoclopramide & ↓ by atropine) affects absorption of other drugs.
- Drugs affecting gut pH:** Gastric acidity ↑ absorption of acidic drugs as aspirin & Intestinal alkalinity ↑ absorption of basic drugs as ephedrine & quinine.
 - Antacids such as calcium carbonate **inhibits** oral absorption of iron preparations.
- Drug binding, adsorption or chelation:** Tetracyclines chelate metals (Ca, Mg, Al & Fe) and decrease their absorption.

- Drugs with **high affinity** to plasma proteins (eg; aspirin & sulfa) can displace other drugs bound to plasma proteins (eg; warfarin) leading to ↑ free concentration and toxicity.

- Enzyme Induction:** HME Inducers **increase** metabolism of the inducer & co-administered drugs; eg: phenobarbitone, phenytoin & tobacco smoking.
- Enzyme Inhibition:** HME Inhibitors **decrease** metabolism of their own & co-administered drugs; eg: chloramphenicol & estrogen.

- Competition for active renal tubular secretion:**
 - Probenecid blocks excretion of weak acids; penicillins.
- Changes in urinary pH:**
 - Acidification** of urine by ammonium chloride ↑ excretion of basic drugs as ephedrine.
 - Alkalinization** of urine by sodium bicarbonate ↑ excretion of acidic drugs as aspirin.

Pharmacodynamic Drug Interactions

1 Interactions at Specific Receptor Sites:
(Pharmacological Antagonism → 2 Drugs 1 Receptor)

Types of Pharmacological Antagonism

A. Competitive Antagonism
(Antagonism displaced by excess agonist)

Example: Acetylcholine & Atropine (Muscarinic Receptor)
Reversible
Block end by Metabolism of Antagonist
eg: Succinylcholine

B. Non-Competitive Antagonism
(Antagonism not displaced by excess agonist)

Irreversible
Bind Covalently, block end by resynthesis of receptor
eg: Phenoxybenzamine

2 Interactions on the Same Physiological System:
Caffeine antagonizes CNS Depressant effect of barbiturates

Interactions Involving Combined Toxicity:

3 Combined use of 2 or more drugs with toxic effects on the same organ can greatly increase organ damage.
Example: Use of 2 nephrotoxic drugs (eg; Aminoglycosides & Cephaloridine)

Types of Antagonism

1 Chemical Antagonism
One drug reacts chemically with an active drug forming an inactive compound.
Example: Neutralization: HCl + NaHCO3.

2 Physiological Antagonism
Two agonists acting on 2 different receptors producing opposite actions
Example: Adrenaline (β2) → bronchodilatation + Histamine (H1) → bronchospasm

3 Pharmacological Antagonism
Two different drugs acting on 1 receptor
Types: Competitive / Non-Competitive

Core Drug Doses

- Therapeutic Dose:** Average dose required to produce therapeutic effect (Adult male: 20-60yrs, 70kgs)
- Loading Dose (LD):** Initial large dose to build up therapeutic blood level
- Maintenance Dose (MD):** Repeated doses at regular intervals to maintain therapeutic blood level
- Minimal Effective Dose:** Lowest dose to produce therapeutic effect
- Maximal Tolerated Dose:** Largest dose without toxic effect.

THERAPEUTIC WINDOW

The dosage range between the minimum effective dose and the minimum toxic dose.

Therapeutic Index
 $TI = LD_{50} / ED_{50}$
LD50 = Dose lethal in 50% Cases
ED50 = Dose effective in 50% Cases
TI is a measure of a drugs safety. Higher TI indicated wide margin between effective and lethal doses → Safer drug

Median Lethal Dose

LD_{50} = Fatal dose in 50% of Cases (Kills 50% of Animals)

Uses of LD_{50}

- Measure of acute Toxicity (Smaller LD_{50} = More Toxic)
- Determine therapeutic index and safety margins
- Clinical trials usually start with 1/10 LD_{50}

Key Formulas

$LD = Vd \times C_{ss}$

$MD = Cl \times C_{ss} \times Tm$

VD: Volume of Distribution
Css: Desired Plasma Conc.
Cl: Clearance **Tm:** Dosing Interval